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Chapter 15 Water and

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SECTION 15.1 WATER

AND ITS PROPERTIES

(pages 445–449) This

section describes the

properties of water in

the liquid and solid

states and explains

how hydrogen bonding

affects the surface

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tension and vapor
pressure of water.

Water in the Liquid
State (pages 445-447)

1.

SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449)

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Chapter 15 "Water and
Aqueous Systems".

The Water Molecule: a
Review. • Water is a

simple tri-atomic
molecule, H₂O.

• Each O-H bond is
highly polar, because
of the high

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electronegativity of the oxygen (N, O, F, and Cl have high values)

- bond angle of water = 105° .

Chapter 15 Water and Aqueous Systems

Chemistry, Chapter 15,
Water and Aqueous
Systems. surface
tension. surfactant.
aqueous solution.
solvent. the inward
force or pull that tends
to minimize the surface

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ar.... any substance
that interferes with

hydrogen bonding

between wa.... is water

that contains dissolved

substances.

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Chapter 15 "Water and Aqueous Systems". use these activities to yourself study the vocabulary and major concepts presented in this chapter.

heterogeneous mixture containing particles that are small enough to remain dispersed in

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the solvent and do not
separate on standing.

Problem

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1. Chapter 15 Chapter
17 in your books“Water
and Aqueous Systems”

2. Section 15.1 Water
and its

PropertiesOBJECTIVES:

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- Explain the high surface tension and low vapor pressure of water in terms of the structure of the water molecule and hydrogen bonding. •

Chemistry - Chp 15 - Water and Aqueous Systems - Notes

Chapter 15 - Water and
Aqueous Systems -

15.3 Heterogeneous

Aqueous Systems -

15.3 Lesson Check -

Page 507: 21 Answer

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Colloids cannot be separated through filtering because their particles are smaller than those of suspensions and they do not settle out over time.

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Contents. Because we want to offer everything you need in one authentic in addition to efficient resource, we all existing beneficial facts about different subject matter plus topics.

Chapter 15 Water And Aqueous Systems Worksheet Answers ...

Title: Chapter 15
Review Water and
Aqueous Systems 1

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Chapter 15 Review
Water and Aqueous
Systems. Pre-AP
Problem

Chemistry ; Charles
Page High School ;
Stephen L. Cotton; 2
Chapter 15 Review.

Surface tension is the
____. How does the
surface tension of
water compare with
the surface tensions of
most other liquids?
Which type of
mixture(s) exhibit the
...

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AP Chemistry Charles
Page High School
Stephen L. Cotton 2.

Section 15.1 Water and
it's Properties □

OBJECTIVES: -Explain
the high surface
tension and low vapor
pressure of water in
terms of the structure
of the water molecule
and hydrogen bonding.

3.

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Chapter 15 "Water and Aqueous Systems" Tools. Copy this to my account; E-mail to a friend; Find other activities ... a relatively strong intermolecular force responsible for water's high surface tension: suspension: a mixture from which particles settle out

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upon standing:

aqueous solution: a

solution in which the
solvent is water:

solvent: the ...

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Chapter 15 - Water and
Aqueous Systems -

15.1 Water and Its
Properties - 15.1

Lesson Check - Page

493: 2 Answer The

structure of ice is a

regular open

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framework of water
molecules arranged
like a honeycomb.

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SECTION 15.1 WATER

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(pages 445-449)

CHAPTER 15, Water and Aqueous

Systems(continued) 6

Circle the letter next to each sentence that describes a result of the surface tension of water

a In a full glass of water, the water surface seems to bulge over the rim of the glass

b Water beads up into

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**17 Water And
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Answers

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