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Chapter 15 Water And Aqueous

Chapter 15 Water and Aqueous
Systems159 SECTION 15.1 WATER AND
ITS PROPERTIES (pages 445–449) This
section describes the properties of water
in the liquid and solid states and
explains how hydrogen bonding affects
the surface tension and vapor pressure

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of water. Water in the Liquid State
(pages 445-447) 1.

SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449)

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Chapter 15 Water and Aqueous Systems.
Chapter 15 "Water and Aqueous Systems". The Water Molecule: a Review. • Water is a simple tri-atomic molecule, H₂O. • Each O-H bond is

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highly polar, because of the high electronegativity of the oxygen (N, O, F, and Cl have high values) • bond angle of water = 105° .

Chapter 15 Water and Aqueous Systems

Chemistry, Chapter 15, Water and Aqueous Systems. surface tension.

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surfactant. aqueous solution. solvent.
the inward force or pull that tends to
minimize the surface ar.... any
substance that interferes with hydrogen
bonding between wa.... is water that
contains dissolved substances.

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yourself study the vocabulary and major concepts presented in this chapter. heterogeneous mixture containing particles that are small enough to remain dispersed in the solvent and do not separate on standing.

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Section 15.1 - Water and its Properties:

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1. Chapter 15 Chapter 17 in your books“Water and Aqueous Systems” 2. Section 15.1 Water and its Properties

OBJECTIVES:• Explain the high surface tension and low vapor pressure

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of water in terms of the structure of the water molecule and hydrogen bonding. •

Chemistry - Chp 15 - Water and Aqueous Systems - Notes

Chapter 15 - Water and Aqueous Systems - 15.3 Heterogeneous Aqueous Systems - 15.3 Lesson Check - Page 507: 21 Answer Colloids cannot be separated

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through filtering because their particles are smaller than those of suspensions and they do not settle out over time.

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Water and Aqueous Systems. Pre-AP
Chemistry ; Charles Page High School ;
Stephen L. Cotton; 2 Chapter 15 Review.
Surface tension is the _____. How does the
surface tension of water compare with
the surface tensions of most other
liquids? Which type of mixture(s) exhibit
the ...

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Chapter 15 "Water and Aqueous Systems" Pre-AP Chemistry Charles Page High School Stephen L. Cotton 2. Section 15.1 Water and it's Properties□

OBJECTIVES: -Explain the high surface tension and low vapor pressure of water in terms of the structure of the water molecule and hydrogen bonding. 3.

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Chapter 15 "Water and Aqueous Systems" Tools. Copy this to my account; E-mail to a friend; Find other activities ... a relatively strong intermolecular force responsible for water's high surface tension:

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suspension: a mixture from which particles settle out upon standing:
aqueous solution: a solution in which the solvent is water: solvent: the ...

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Chapter 15 - Water and Aqueous Systems - 15.1 Water and Its Properties -

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15.1 Lesson Check - Page 493: 2 Answer
The structure of ice is a regular open framework of water molecules arranged like a honeycomb.

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Systems - 15.2 Homogeneous Aqueous
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SECTION 15.1 WATER AND ITS
PROPERTIES (pages 445-449) CHAPTER
15, Water and Aqueous

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Systems(continued) 6 Circle the letter next to each sentence that describes a result of the surface tension of water a In a full glass of water, the water surface seems to bulge over the rim of the glass b Water beads up into

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