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0.35 «cm³ s⁻¹» Accept values in the range 0.32–0.42 «cm³ s⁻¹». 3 2. b ALTERNATIVE 1 colour Br₂ /reactant is coloured «Br (aq) is not» ALTERNATIVE 2 conductivity greater/increased concentration of ions in products ALTERNATIVE 3 mass/pressure gas is evolved/produced ALTERNATIVE 4 pH

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1.3 10 76 10 » 1.7 103/1711 ALTERNATIVE 2 1600 76 106 1.2 1011 «is less than tensile strength of graphene» ALTERNATIVE 3 1.3 1011 1600 8.1 107 «is greater than upper end of tensile strength for graphite» Accept any value in the range 1700–27083. Answer may be expressed in scientific notation or otherwise.

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hl CHEMISTRY NOTES. TOPIC 1: QUANTITATIVE CHEMISTRY. TOPIC 2: ATOMIC STRUCTURE ... 9: REDOX PROCESSES. TOPIC 10: ORGANIC CHEMISTRY. TOPIC 11: MEASUREMENT AND DATA PROCESSING. OPTION B: BIOCHEMISTRY. THE HELPFUL CHEMISTRY RESOURCES I FOUND USEFUL. RadioChemistry: IB Online Teacher; ... past papers. specimen papers. Powered by Create your own ...

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Subject Details: Chemistry HL Paper 3 Markscheme Mark Allocation Candidates are required to answer questions from TWO of the options [2 x 25 marks]. Maximum total = [50 marks]. 1. A markscheme often has more marking points than the total allows. This is intentional. Do not award more than the maximum marks allowed for part of a question. 2.

Higher Level Paper 3 - Xtreme

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The 1:2 ratio of marks (paper 3-to-paper 2) for SL versus the 2:3 ratio for HL is important to bear in mind when deciding how much time to give over to each exam in preparation. On a similar note, the emphasis on the Option is different for SL and HL. The Option forms a much smaller fraction of the overall mark if you are studying it at SL.

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